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# Writing And Naming Binary Compounds Worksheet Answer Key

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*Naming Ionic and*

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---

Writing Ionic

Formulas:

Introduction **Naming**

**Ionic Compounds**

*Naming Ionic*

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*Introduction Writing*

*Ionic Formulas - Basic*

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*Binary Compounds*

*How to name and*

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*Name Ionic*

*Compounds With*

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*Formulas Lesson 1:*

*Writing Formulas For*

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Writing And

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Compounds Naming

Compounds with

Polyatomic Ions Lewis

Diagrams Made Easy:

How to Draw Lewis

Dot Structures

*VSEPR Theory:*

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Naming Ionic

Compounds with

Roman Numerals!

---

Naming Compounds

in Chemistry *Chemical*

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Writing And

*Formulas - Explained  
Formulas and Names  
of Molecular  
Compounds*

---

**Naming Ionic  
Compounds Tricks for  
Remembering  
Polyatomic Ions**

**Periodic Trends:  
Electronegativity,  
Ionization Energy,  
Atomic Radius -  
TUTOR HOTLINE  
Valence Electrons**



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**Naming Binary**  
**Table Naming**  
Covalent Molecular  
Compounds *Naming*  
*Type 1 Binary*  
*Compounds Part 1*  
Nomenclature  
(~~Writing and Naming~~  
~~Binary Ionic~~  
~~Compounds~~) ~~Type III~~  
~~Binary Compounds~~  
~~Naming and Writing~~  
~~Formulas~~ Type II  
Binary Ionic

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Writing And

Compounds - Naming  
and Writing Formulas

Naming Ionic

Compounds! (Simple  
Binary Ionic) 10-5.1

and 5.2 Naming

Compounds; Naming

Binary Compounds

That Contain a Metal

and a Nonmetal *HOW*

*TO NAME TYPE II*

*IONIC COMPOUNDS*

**Writing And Naming**

**Binary Compounds**

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Writing And

WRITING AND

NAMING BINARY

COMPOUNDS

WORKSHEET. WKS

7.1 – Beginning

Naming & Formula

Writing (2 pages)

Name the following  
ionic compounds. Be

sure to check for

Roman Numerals!

Na<sub>3</sub>P. MgBr<sub>2</sub>. Ag<sub>2</sub>Se.

AlCl<sub>3</sub>. Ca<sub>3</sub>P<sub>2</sub>.

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Writing And

**WRITING AND  
NAMING BINARY  
COMPOUNDS  
WORKSHEET**

WRITING AND

NAMING BINARY

COMPOUNDS Name

\_Joshua

Roberts\_\_\_\_\_ Date

\_\_11/8/2020\_\_\_\_\_

Period \_5\_\_ Write a

correct formula for

each of the following

binary compounds:

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Naming Binary

          $\text{Cs}_2\text{S}$           1.  
cesium sulfide

          $\text{Ba}_3\text{N}_2$          

11. barium nitride

          $\text{ZnO}$           2.

zinc oxide

          $\text{Li}_2\text{O}$           12.

lithium oxide         

          $\text{CdF}_2$           3.

cadmium fluoride

          $\text{KCl}$           13.

potassium chloride

          $\text{AlI}_3$           4.

aluminum iodide

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CaBr<sub>2</sub>

14. calcium bromide

AgNO<sub>3</sub>

5. silver nitride

Na<sub>2</sub>S 15.

sodium ...

**Writing and Naming  
Binary Compounds  
.pdf - WRITING AND**

...

Binary covalent  
compounds are  
compounds made up

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Naming Binary Compounds  
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Answer Key

of only two elements, such as carbon dioxide. Prefixes are used in the names of binary compounds to indicate the number of atoms of each nonmetal present.

The following table lists the most common prefixes for binary covalent compounds. In general, the prefix

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mono- is rarely used.

Compounds

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Key Cadmium Sulfide.

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Naming Covalent

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Binary Ionic

Compounds

Polyatomic

Ions Naming with

Polyatomic Ions

Naming with Roman

Numerals Formula

Writing Naming Acids.

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Writing And

Naming Binary Ionic  
Compounds. Rules. 1.  
The Cation (positive  
ion) is named first, the  
Anion second. 2.

Monoatomic Cations  
take the element  
name.  $\text{Na}^{+}$  -->

Sodium.  $\text{Ca}^{2+}$  -->  
Calcium.

**Naming Binary Ionic  
Compounds -  
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Writing And

Writing Formulas for

Binary Ionic

Compounds Learning

to Name and Write

Formulas: - Key

Determining the type of compound you are working with. -

Applying the rules for naming or formula writing for that type of compound.

**Writing Formulas for**

*Page 21/37*

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**BINARY IONIC**

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Writing And

COMPOUNDS

Writing Formulas from  
Names • 1st word =

CATION • 2nd word =  
ANION name with ide  
ending. BINARY

IONIC COMPOUNDS

NaBr MgF<sub>2</sub> Sodium

Bromide Magnesium

Fluoride BINARY

IONIC COMPOUNDS

Potassium Chloride

Aluminum Oxide •

notice ending of name

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Writing And  
is ide! Binary

Compounds

**PowerPoint**

**Presentation -**

**Chemical Names  
and Formulas**

Compound Name

Formula Search »

Moles to Grams

Calculator » Common

Compounds List »

Chemical Equation

Balancer » Complete

List of Acids »



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Complete List of  
Bases » Molar to  
Mass Concentration  
Converter » Molar  
Mass Calculator »  
Cations, Anions List »  
Dilution Calculator »  
Molarity Calculator »  
Compound Prefixes »  
Water Insoluble ...

**Compound Name**  
**Formula Search --**  
**EndMemo**

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Writing And

deca-. The rules for using the prefix system of nomenclature of binary compounds can be summarized as follows. Generally, the less-electronegative element is written first in the formula, though there are a few exceptions. Carbon is always first in a

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Writing And

formula and hydrogen  
is after nitrogen in a  
formula such as  $\text{NH}_3$ .

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## **7.11: Naming Binary Molecular**

**Compounds -**

**Chemistry**

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You learned that  
naming simple binary  
ionic compounds is  
easy. Name the  
cation first with the

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Writing And

element name, then

name the anion and

add -ide to the end.

Naming ionic

compounds with

transition metals isn't

too hard either. They

are named like the

binary compounds,

with the cation first,

then the anion with

-ide added to it, but

you have to take into

account the variations

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of the metal ions.

Compounds  
**Naming Ionic  
Compounds: Simple  
Binary, Transition  
Metal ...**

Can you write  
formulas for binary  
ionic compounds?  
Can you name binary  
ionic compounds?  
Let's find out...

**Quia - Binary Ionic**  
*Page 29/37*

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Writing And

## Compounds

OBJ: 9.2.1 Apply the rules for naming and writing formulas for binary ionic

compounds. STA:

Ch.2 18.ANS: C PTS:

1 DIF: L2 REF: p. 257

| p. 261 | p. 262 OBJ:

9.2.2 Apply the rules for naming and writing formulas for

compounds with

polyatomic ions. STA:

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Writing And

Ch.2 19.ANS: C PTS:

1 DIF: L2 REF: p. 264

| p. 277

Worksheet

**Chapter 9 Practice**

**Test - Naming and**

**Writing Chemical**

**Formulas**

Start studying

Naming: Binary Ionic

Compound Practice.

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terms, and more with

flashcards, games,

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and other study tools.

Compounds  
**Naming: Binary  
Ionic Compound  
Practice Flashcards  
| Quizlet**

Get more practice  
naming compounds.  
For Al<sub>2</sub>O<sub>3</sub> use the  
hints and resources  
below to help write  
the name. Hint for  
Naming Al<sub>2</sub>O<sub>3</sub>.

Aluminum (Al) is a



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Writing And

metal. Oxygen (O) is a non-metal. This is a ionic compound. First name the metal as found on the Periodic Table. Next name the non-metal (Ox ygen) and change the ygen ending to ide .

**How to Write the Name for  $Al_2O_3$  - The Geoexchange**

For binary ionic

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Compounds (ionic compounds that contain only two types of elements), the compounds are named by writing the name of the cation first followed by the name of the anion. For example,  $\text{KCl}$ , an ionic compound that contains  $\text{K}^+$  and  $\text{Cl}^-$  ions, is named potassium chloride.

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Naming ions and  
ionic compounds  
(video) | Khan  
Academy Key

Name: Ranesh Verma

Per:5 Date: 11/30/20

Chemistry: Covalent

Binary Compounds:

Nonmetal–Nonmetal

Combinations Write

the name of each of

the following

compounds. 1. HF 1.

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Hydrogen fluoride 2.

$H_2S_2$ . Hydrogen

Sulfide 3.  $NO_3$ . Nitric

oxide 4.  $N_2O_4$ .

Nitrous oxide 5.  $NO_2$

5. Nitrogen Dioxide 6.

$N_2O_5$  6. Dinitrogen

pentoxide 7.  $SO_2$  7.

Sulfur ...

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7515e1d0da

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